

Addition/Correction

Zinc–Zinc Bonded Zincocene Structures. Synthesis and Characterization of Zn(η -CMe) and Zn(η -CMeEt) [*J. Am. Chem. Soc.* 2007, 129, 693–703].

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J. Am. Chem. Soc., 2007, 129 (45), 14100-14100 • DOI: 10.1021/ja075362m • Publication Date (Web): 18 October 2007

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Zinc–Zinc Bonded Zincocene Structures. Synthesis and Characterization of $\text{Zn}_2(\eta^5\text{-C}_5\text{Me}_5)_2$ and $\text{Zn}_2(\eta^5\text{-C}_5\text{Me}_4\text{Et})_2$ [*J. Am. Chem. Soc.* **2007**, *129*, 693–703]. Abdessamad Grrirane, Irene Resa, Amor Rodríguez, Ernesto Carmona,* Eleuterio Alvarez, Enrique Gutiérrez-Puebla, Angeles Monge, Agustín Galindo, Diego del Río, and Richard A. Andersen

Page 702. The synthesis of compound **2**, $\text{Zn}_2(\eta^5\text{-C}_5\text{Me}_4\text{Et})_2$, should be as follows:

Compound 2, $\text{Zn}_2(\eta^5\text{-C}_5\text{Me}_4\text{Et})_2$. A mixture of $\text{KC}_5\text{Me}_4\text{Et}$ (3.10 g, 36.6 mmol) and ZnCl_2 (2.5 g, 18.6 mmol) was dissolved in THF (50 mL) and stirred for 4 h at room temperature. The solvent was evaporated in vacuo, and the residue was extracted with pentane (3 × 15 mL). After removing the pentane, the Zn(II) metallocene $\text{Zn}(\text{C}_5\text{Me}_4\text{Et})_2$ was obtained as a yellow oil in ca. 70% yield. Reduction of $\text{Zn}(\text{C}_5\text{Me}_4\text{Et})_2$ as above ($\text{Zn}(\text{C}_5\text{Me}_4\text{Et})_2$ (2.2 g, 6.2 mmol), ZnCl_2 (840 mg, 6.2 mmol), and KH (494 mg, 12.3 mmol) in THF (30 mL)) for 50 min at 10–20 °C and evaporation of the volatiles in vacuo gave a residue that was dissolved in 40 mL of pentane and filtered. It is advisable to perform these operations at low temperatures (0 to –10 °C). Crystals of **2** suitable for X-ray studies were obtained by slow evaporation of its pentane solutions at –20 °C or by low-temperature (–80 °C) crystallization of concentrated pentane solutions. Compound **2** is also highly reactive toward O_2 and H_2O and has lower thermal stability than **1** decomposing slowly, particularly in the form of a solid, at temperatures between 0 and 20 °C. It is advisable to store it at –20 °C (or below) under argon. ^1H NMR (500 MHz, C_6D_6): δ 1.06 (t, $J_{\text{HH}} = 7.5$ Hz, 3H, $\text{CH}_3\text{-Et}$), 1.99 (s, 6H, $\text{CH}_3\text{-Cp}$), 2.00 (s, 6H, $\text{CH}_3\text{-Cp}$), 2.43 (q, $J_{\text{HH}} = 7.5$ Hz, 2H, $\text{CH}_2\text{-Et}$). $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, C_6D_6): δ 9.7 (s, $\text{CH}_3\text{-Cp}$), 9.9 (s, $\text{CH}_3\text{-Cp}$), 18.1 (s, $\text{CH}_3\text{-Et}$), 18.5 (s, $\text{CH}_2\text{-Et}$), 107.9 (s, $\text{C}_q\text{-Cp}$), 108.2 (s, $\text{C}_q\text{-Cp}$), 116.9 (s, $\text{C}_q\text{-Cp}$). Yield: 36% (950 mg).

JA075362M

10.1021/ja075362m

Published on Web 10/18/2007

Palladium-Catalyzed Direct Functionalization of Imidazolinone: Synthesis of Dibromophakellstatin [*J. Am. Chem. Soc.* **2007**, *129*, 7768–7769]. Jianming Lu, Xianghui Tan, and Chuo Chen*

Page 7769. The references for the C–H insertion mechanism were inadvertently omitted:

References

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JA077266R

10.1021/ja077266r

Published on Web 10/23/2007